PRE-COMMERCIALISATION DEVELOPMENT OF COPPER REDOX FLOW BATTERIES

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ABSTRACT

Copper redox flow batteries (CuRFB) provide a desired middle ground between the high raw material cost of VRFB and lower energy efficiency of iron flow batteries. Recent electrolyte and system design reserach has significantly improved CuRFB power & energy density, resulting in a highly scalable and low-cost ownership battery technology for adopters with a levelized cost of storage of 75 €/MWh over the projected lifespan of 20 years. During the CuRen project system size has increased dramatically, from 0.5 W to 200 W, with a planned 1 kW system soon to be completed.

COPPER RFB CHEMISTRY

CuRFB uses a symetrical electrolyte, cycling via disproportionation & comproportionation reactions, with deposition occuring at the cathode. Halides are added to solution to stabalise the copper active species, forming copper complexes. Lower cell voltages enables the use of a acidic environment, increasing conductivity.

OPERATIONAL ADVANTAGES

Simple chemistry enables simple design with several all-round improvements to operation.

CU-HALIDE ELECTROLYTES

The stabalisation of copper by bromide increases power density by 45%, significantly reducing the required system size.



Cycling is uneffected by impurities in the electrolyte, with the use of 97% pure copper sources demonstrating no loss in performance. This allows for the future use of second-life copper in a single-step electrolyte production process.

Dendrite growth is non-apparent, with no risk of precipitation and cell death. Despite the lack of a highly oxidative species, the high acidity necessitates specific materials choice. High cycling stability and cell performance have enabled CuRFB as a competivive 4-12 hour storage alternative.

CE (%)	>98	EE (%)	75
Dowor donaity		Energy density	

- **High stability:** 100% Depth of charge and fully recoverable capacity².
- Improved safety: No H2/O2 evolution, large ambient operating window.
- Sustainable: Zero PGM usage, > 99% Copper recovery end of life, majority recyclable system.³





The improved electrolyte decreases Deployed System LCOS to 74 €/MWh, where battery LCOS is 45 €/MWh





Expanded view of a CuRen V3 stack.

V: Vanadium RFB US DOE $V^1 \boxtimes \text{Best V} \otimes \text{CuCl} \boxtimes \text{CuBr}$

TOWARDS A 1 KW STACK

Alongside increasing stack size, other material and electrolyte improvements have been pursued. These are to be exploited in a 1 kW stack by Q3 2024.

- Custom fabricated SPEEK-biochar composite membranes enable minimal capacity decay at low cost.
- Novel stack sealing methods removes traditional gaskets, decreasing cell width.
- The higher solubility of copper bromide complexes in low pH enables a simple 3 molar copper electrolyte.

Project system development:



COMMERCIAL DEVELOPMENT

CuRen aims to spin-out by Q4 2024 as a CuRFB manufacturer and supplier based in Finland with a complete EU supply chain. System plans include containerised battery systems (120 kW, 960 kWh) delivered with an integrated battery management system, and continuous operation & maintenance services.

Current IP includes:

- Copper halide systems
- CuRFB integrated rebalancing system
- Novel low cost stack & system manufacturing

Power (W)



CONCLUSION

Research into CuRFBs has yielded significant increases in the systems' energy density drastically reducing battery LCOS to well below the DOE target of < €50/MWh and that of existing redox flow battery chemistries. The high energy and low environmental impact system provideds an all-round high performance energy storage solution, which has enabled steady progress towards commercialisation. However, there is yet much interest in both further engineering and reserach work towards a more sustainable, cost-effective and power dense system.

CU REN

Modular CuRFB unit (3 kW, 24 kWh), 40 units in a containerised system for 120 kW, 960 kWh.

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US Department of Energy: Grid Energy Storage Technology Cost and Performance Assessment (2022), PNNL-33283
Badenhorst, Wouter Dirk, et al., "Performance improvements for the all-copper redox flow battery: Membranes, electrodes, and electrolytes." Energy Reports 8 (2022): pp. 8690-8700.
Lloyd, David, et al. "Preparation of a cost-effective, scalable and energy efficient all-copper redox flow battery." Journal of Power Sources 292 (2015): 87-94.

